The same will be true for any one-electron operator, so for the electron-nuclear attraction

$$
\left.<-\frac{2}{r_{1 n}}>=\left\langle-\frac{2}{r_{2 n}}\right\rangle=2<-\frac{2}{r_{1 n}}\right\rangle
$$

The 2-electron operator $1 / r_{12}$ is different:

$$
<\frac{1}{r_{12}}>=\int 1 s(1) 1 s(2) \frac{1}{r_{12}} 1 s(1) 1 s(2) d \tau_{1} d \tau_{2}=J_{1 s, 1 s}
$$

$\mathrm{J}_{\mathrm{ij}}$ repulsion integrals are called "Coulomb integrals";
Here they are simply the repulsion of two 1s clouds of negative charge, that happen to be superimposed in this case.

Thus, for the He ground state:
$E_{\text {ground }}=\mathbf{2} h_{1 \mathrm{~s}}+J_{1 \mathrm{~s}, 1 \mathrm{~s}}$
where $h_{1 s}=$ is a sum of one-electron operators:
the KE + electron-nuclear attraction for a 1s electron

## Next consider the Helium 1s2s configuration

$$
\begin{gathered}
\Psi_{\text {singlet }}=[1 s(1) 2 s(2)+2 s(1) 1 s(2)][\alpha(1) \beta(2)-\beta(1) \alpha(2)] \\
\Psi_{\text {singlet }}=[1 s(1) 2 s(2)+2 s(1) 1 s(2)]\{\alpha(1) \beta(2)-\beta(1) \alpha(2)\} \\
\langle E\rangle_{\text {singlee }}=[<1 s(1) 2 s(2)+2 s(1) 1 s(2)]\{\alpha(1) \beta(2)-\beta(1) \alpha(2)\}|H|[1 s(1) 2 s(2)+2 s(1) 1 s(2)]\{\alpha(1 \\
\left.\langle E\rangle_{\text {singlet }}=[11 s(1) 2 s(2)+2 s(1) 1 s(2)]\{\alpha(1) \beta(2)-\beta(1) \alpha(2)\}|H|[1 s(1) 2 s(2)+2 s(1) 1 s(2)]\{\alpha(1) \beta(2)-\beta(1) \alpha(2)\}\right\rangle
\end{gathered}
$$

Again, H does not have spin coordinates , ignoring spin-orbit coupling, so:
$E_{\text {singlet }}=\langle s(1) 2 s(2)+2 s(1) 1 s(2)| H|s(1) 2 s(2)+2 s(1) 1 s(2)\rangle$
Now the 1-electron part has to be $\mathbf{h}_{1 s} \mathbf{+} \mathbf{h}_{\mathbf{2 s}}$ but appears to be twice that at this point. Must normalize the linear combination,

## Normalization

Whenever two orthonormal functions are added, e.g., $|A\rangle+|B\rangle$, the overlap is no longer 1 , but $(\langle A|+\langle B|)|(|A\rangle+|B\rangle)$

$$
\begin{aligned}
& =N^{2}(\langle A \mid A\rangle+\langle B \mid B\rangle+\langle A \mid B\rangle+\langle B \mid A\rangle \\
& =1+1+0+0=2
\end{aligned}
$$

So normalization constant squared $=1 / 2$ norm const $=\mathbf{2}^{-1 / 2}=0.707$
Generally, adding N orthonormal functions with equal weight gives norm const $=\mathrm{N}^{-1 / 2}$
but the 2-electron part has an extra part due to electron exchange

$$
\begin{aligned}
& \left.<1 / r_{12}\right\rangle=\frac{1}{2}\langle 1 s(1) 2 s(2)+2 s(1) 1 s(2)| \frac{1}{r_{12}}|1 s(1) 2 s(2)+2 s(1) 1 s(2)\rangle \\
& \left.=\langle 1 s(1) 2 s(2)| \frac{1}{r_{12}}|1 s(1) 2 s(2)\rangle+\langle 1 s(1) 2 s(2)| \frac{1}{r_{12}} \right\rvert\, \begin{array}{c}
\text { exchanged } 1 \text { and } 2 \\
2 s(1) 1 s(2)\rangle
\end{array} \\
& =\langle 1 s(1) 1 s(1)| \frac{1}{r_{12}}|2 s(2) 2 s(2)\rangle+\langle 1 s(1) 2 s(1)| \frac{1}{r_{12}}|1 s(2) 2 s(2)\rangle
\end{aligned}
$$

Order does not matter (just multiplying by $1 / r_{12}$ ). Writing this way makes it easier to visualize

$$
=\quad J_{1 s, 2 \mathrm{~s}} \quad+\quad \mathrm{K}_{1 \mathrm{~s}, 2 \mathrm{~s}}
$$

where $\mathbf{K}_{1 s, 2 s}$ is called an exchange integral because it arises only because of the requirement that the electrons must exchange.
$\mathrm{J}_{\mathrm{ij}}$ is classical; It is simply the Coulombic repulsion of $1 \mathrm{~s}(1)^{2}$ and $2 \mathrm{~s}(2)^{2}$ negative clouds of charge.
$\mathrm{K}_{\mathrm{ij}}$ is the repulsion of two ij clouds: $1 \mathrm{~s}(1) 2 \mathrm{~s}(1)$ and $1 \mathrm{~s}(2) 2 \mathrm{~s}(2)$

## Triplet State

$\left.<1 / r_{12}\right\rangle=\frac{1}{2}\langle 1 s(1) 2 s(2)-2 s(1) 1 s(2)| \frac{1}{r_{12}}|1 s(1) 2 s(2)-2 s(1) 1 s(2)\rangle$
$=\langle 1 s(1) 2 s(2)| \frac{1}{r_{12}}|1 s(1) 2 s(2)\rangle-\langle 1 s(1) 2 s(2)| \frac{1}{r_{12}}|2 s(1) 1 s(2)\rangle$
$=\langle 1 s(1) 1 s(1)| \frac{1}{r_{12}}|2 s(2) 2 s(2)\rangle-\langle 1 s(1) 2 s(1)| \frac{1}{r_{12}}|1 s(2) 2 s(2)\rangle$
Now the cross terms are negative, because the space function is antisymmetric.
$=\quad \mathrm{J}_{1 \mathrm{~s}, 2 \mathrm{~s}}$
minus
$\mathrm{K}_{1 \mathrm{~s}, 2 \mathrm{~s}}$

The triplet state is LOWER than the singlet state!!! because it can be proved that all $\mathrm{I}_{\mathrm{ij}}$ and $\mathrm{K}_{\mathrm{ij}}$ integrals are positive, but $\mathrm{K}_{\mathrm{ij}}<\mathrm{J}_{\mathrm{ij}}$.
$\mathrm{J}_{\mathrm{ij}}$ is classical; It is simply the Coulombic repulsion of two negative clouds of charge described by the squares of the orbitals $\mathrm{K}_{\mathrm{ij}}$ is the repulsion of two ij clouds e.g., $1 \mathrm{~s}(1) 2 \mathrm{~s}(1)$

## Slater Determinants notation for the same functions:

$$
\begin{gathered}
\Psi_{\text {ground }}=\varphi_{1 s}(1) \bar{\varphi}_{1 s}(2)-\bar{\varphi}_{1 s}(1) \varphi_{1 s}(2) \\
\Psi_{\text {ground }}=\left|\begin{array}{ll}
\varphi_{1 s}(1) & \bar{\varphi}_{1 s}(1) \\
\varphi_{1 s}(2) & \bar{\varphi}_{1 s}(2)
\end{array}\right| \\
\Psi=A(1) B(2)-B(1) A(2) \\
\Psi_{\text {ground }}=\left|\begin{array}{ll}
A(1) & B(1) \\
A(2) & B(2)
\end{array}\right|=\left|\begin{array}{cc}
A(1) & A(2) \\
B(1) & B(2)
\end{array}\right| \equiv|A B|
\end{gathered}
$$

For 4 electrons:
$|A B C D|$ means a a slater determinant and the electron numbering in each term is $1,2,3,4$

## Antisymmetry for more than 2 electrons

## Generic 3 electron case:

$$
\begin{aligned}
& \Psi=A(1) B(2) C(3) \ldots-B(1) A(2) C(3) \ldots-C(1) B(2) A(3) \ldots \\
& \Psi=\left|\begin{array}{lll}
A(1) & A(2) & A(3) \\
B(1) & B(2) & B(3) \\
C(1) & C(2) & C(3)
\end{array}\right| \equiv\left|\begin{array}{ccc}
A(1) & B(1) & C(1) \\
A(2) & B(2) & C(2) \\
A(3) & B(3) & C(3)
\end{array}\right| \equiv|A B C| \\
& \left.\Psi^{*} \Psi=\langle | A B C|\| A B C|\right\rangle
\end{aligned}
$$

Note that it does not matter whether: columns have same electron number and rows have same orbital, or vice versa

## Normalization (assuming orthonormal orbitals)

There are $\mathbf{N}$ ! ( $\mathbf{N}$ factorial) terms for an N -electron Slater determinant
Example of 3 electrons: $3!=6$
$A B C-A C B+B C A-B A C+C A B-C B A$
Overlap integral is:
$\angle A B C-A C B+B C A-B A C+C A B-C B A \mid A B C-A C B+B C A-B A C+C A B-C B A>=?$
$=\angle A B C \mid A B C>=\langle A| A>\angle B|B><C| C\rangle=(1)(1)(1)$ because orbitals are 1 -electron functions
but, if there is only one mismatch, e.g., <ABC|ACB>
$=\langle A \mid A\rangle\langle B \mid C\rangle\langle C \mid B\rangle=(1)(0)(0)=0$

## ab initio Calculations

meaning literally: from the beginning

In practice means: no approximations are made in the mathematical operations, although all calculations are more or less approximate, depending on the number of variational parameters chosen.

$$
H=\sum_{i}-\frac{1}{2} \nabla_{i}^{2}-\sum_{i, \mu} \frac{Z_{\mu}}{r_{i \mu}}+\sum_{i>j} \frac{1}{r_{i j}}+\sum_{i}-\frac{1}{2} \nabla_{\mu}^{2}+\sum_{\mu, v} \frac{Z_{\mu \nu}}{r_{\mu \nu}}
$$

$$
\begin{aligned}
& H=\sum_{i}-\frac{1}{2} \nabla_{i}^{2}-\sum_{i, \mu} \frac{Z_{\mu}}{r_{i \mu}}+\sum_{i>j} \frac{1}{r_{i j}}+\sum_{i}-\frac{1}{2} \nabla_{\mu}^{2}+\sum_{\mu, v} \frac{Z_{\mu v}}{r_{\mu \nu}} \\
& \underbrace{\begin{array}{l}
\text { electron } \\
\text { kinetic E } \\
\text { operators }
\end{array}}_{\text {ONE-electron }} \begin{array}{l}
\begin{array}{l}
\text { electron- } \\
\text { nuclear } \\
\text { attraction } \\
\text { operators }
\end{array} \\
\text { TWO-electron }
\end{array} \\
& \text { nuclear nuclear- } \\
& \text { kinetic E nuclear } \\
& \begin{array}{l}
\text { = } \begin{array}{l}
\text { e for electronic } \\
\text { energy at a fixed } \\
\text { geometry }
\end{array} \\
\text { = a constant }
\end{array} \\
& \text { for electronic } \\
& \text { energy at a fixed } \\
& \text { geometry } \\
& \text { Is added to the } \\
& \text { electronic energy } \\
& \text { at the end }
\end{aligned}
$$

## Hartree Self-Consistent Field (SCF)

Wavefunction $=$ single product of orthonormal spin orbitals

## $A \bar{A} B \bar{B} C \bar{C} \ldots \quad$ where bar on top means $\beta$ spin

This preceded the Hartree-Fock SCF.

What is wrong with this?

Not anti-symmetric to interchange of electron coordinates,
Fock was the one who extend the method for Slater Determinants.

## Hartree-Fock Self Consistent Field (SCF

Wavefunction $=$ single Slater determinant of orthonormal spin orbitals $\quad(N!)^{-1 / 2}|A \bar{A} B \bar{B} C \bar{C} \ldots|$ HF-SCF total energy: ***

## Simplifications:

1. ${ }^{* * *}$ Unnecessary to have a determinant on both sides because of double counting of permutations
2. Integrate over electrons ONLY because :
nuclear kinetic energy assumed zero (Born-Oppenheimer)
3. and therefore nuclear-nuclear repulsion is a constant to be added after the purely electronic energy is determined.
4. H does not contain spin (a good approximation for light atoms)

Consider the example where there are $6!=720$ permutations on the right, showing only the permutation of electrons 1 and 2

$$
\begin{aligned}
& E_{\text {trial }}=(N!)^{-1 / 2}<A \bar{A} B \bar{B} C \bar{C}|H||A \bar{A} B \bar{B} C \bar{C}|> \\
& =<\mathbf{A} \overline{\mathbf{A}} B \bar{B} C \bar{C}|H| \mathbf{A} \overline{\mathbf{A}} B \bar{B} C \bar{C}-\overline{\mathbf{A}} \mathbf{A} B \bar{B} C \bar{C}-\ldots>
\end{aligned}
$$

$$
\langle\mathrm{A}(1)| \mathrm{h}(1)|\mathrm{A}(1)\rangle\langle\mathrm{A}(2) \mid \mathrm{A}(2)\rangle\langle\alpha| \beta><\beta|\alpha\rangle\langle\mathrm{A}(3) \mid \mathrm{A}(3)\rangle\langle\mathrm{A}(4) \mid \mathrm{A}(4)\rangle \ldots .
$$ just for the one-electron part of $\boldsymbol{H}$ for electron 1 .

This illustrates most of the permutations give zero because of either spin or space orthogonality mismatches (when an electron is in a different spin-orbital on the two sides of the integral).

Such mismatches are may only be non zero if the corresponding operator in the Hamiltonian intervenes.

## Slater Condon Rules

Slater and Condon articulated rules for the few cases on non zero element.

### 11.8 The Condon-Slater Rules

In the Hartree-Fock approximation, the wave function of an atom (or molecule) is a Slater determinant or a linear combination of a few Slater determinants [for example, Eq. (10.44)]. A configuration-interaction wave function such as (11.17) is a linear combination of many Slater determinants. To evaluate the energy and other properties of atoms and molecules using Hartree-Fock or configuration-interaction wave functions, we must be able to evaluate integrals of the form $\left\langle D^{\prime}\right| \hat{B}|D\rangle$, where $D$ and $D^{\prime}$ are Slater determinants of orthonormal spin-orbitals and $\hat{B}$ is an operator.

Each spin-orbital $u_{i}$ is a product of a spatial orbital $\theta_{i}$ and a spin function $\sigma_{i}$, where $\sigma_{i}$ is either $\alpha$ or $\beta$. We have $u_{i}=\theta_{i} \sigma_{i}$ and $\left\langle u_{i}(1) \mid u_{j}(1)\right\rangle=\delta_{i j}$, where $\left.\left\langle u_{i}(1)\right| u_{j}(1)\right)$ involves a sum over the spin coordinate of electron 1 and an integration over its spatial coordinates. If $u_{i}$ and $u_{j}$ have different spin functions, then (10.12) ensures the orthogonality
$u_{i}$ and $u_{j}$. If $u_{i}$ and $u_{j}$ have the same spin function, their orthogonality is due to the arthogonality of the spatial orbitals $\theta_{i}$ and $\theta_{j}$.

For an $n$-electron system, $D$ is

$$
D=\frac{1}{\sqrt{n!}}\left|\begin{array}{cccc}
u_{1}(1) & u_{2}(1) & \ldots & u_{n}(1)  \tag{11.76}\\
u_{1}(2) & u_{2}(2) & \ldots & u_{n}(2) \\
\vdots & \vdots & \ddots & \vdots \\
u_{1}(n) & u_{2}(n) & \ldots & u_{n}(n)
\end{array}\right|
$$

An example with $n=3$ is Eq. (10.40). $D^{\prime}$ has the same form as $D$ except that $u_{1}, u_{2}, \ldots, u_{n}$ are replaced by $u_{1}^{\prime}, u_{2}^{\prime}, \ldots, u_{n}^{\prime}$.

If $u_{i}$ and $u_{j}$. If $u_{i}$ and $u_{j}$ have the same spin function, their orthogonality is due to the rthogonality of the spatial orbitals $\theta_{i}$ and $\theta_{j}$.

For an $n$-electron system, $D$ is

$$
D=\frac{1}{\sqrt{n!}}\left|\begin{array}{cccc}
u_{1}(1) & u_{2}(1) & \ldots & u_{n}(1)  \tag{11.76}\\
u_{1}(2) & u_{2}(2) & \ldots & u_{n}(2) \\
\vdots & \vdots & \ddots & \vdots \\
u_{1}(n) & u_{2}(n) & \ldots & u_{n}(n)
\end{array}\right|
$$

An example with $n=3$ is Eq. (10.40). $D^{\prime}$ has the same form as $D$ except that $u_{1}, u_{2}, \ldots, u_{n}$ are replaced by $u_{1}^{\prime}, u_{2}^{\prime}, \ldots, u_{n}^{\prime}$.

We shall assume that the columns of $D$ and $D^{\prime}$ are arranged so as to have as many as possible of their left-hand columns match. For example, if we were working with the Slater determinants $\left|1 s \bar{s} 2 s 3 p_{0}\right|$ and $\left|1 s \overline{1} \bar{s} 3 p_{0} 4 s\right|$, we would interchange the third and fourth columns of the first determinant (thereby multiplying it by -1 ) and let $D=\left|1 s \overline{1} \bar{s} 3 p_{0} 2 s\right|$ and $D^{\prime}=\left|1 s \bar{s} 3 p_{0} 4 s\right|$.

The operator $\hat{B}$ typically has the form

$$
\begin{equation*}
\hat{B}=\sum_{i=1}^{n} \hat{f}_{i}+\sum_{i=1}^{n-1} \sum_{j>i} \hat{g}_{i j} \tag{11.77}
\end{equation*}
$$

where the one-electron operator $\hat{f}_{i}$ involves only coordinate and momentum operators of electron $i$ and the two-electron operator $\hat{g}_{i j}$ involves electrons $i$ and $j$. For example, if $\hat{B}$ is the atomic Hamiltonian operator (11.1), then $\hat{f}_{i}=-\left(\hbar^{2} / 2 m_{e}\right) \nabla_{i}^{2}-Z e^{2} / 4 \pi \varepsilon_{0} r_{i}$ and $\hat{g}_{i j}=e^{2} / 4 \pi \varepsilon_{0} r_{i j}$.

Condon and Slater showed that the $n$-electron integral $\left\langle D^{\prime}\right| \hat{B}|D\rangle$ can be reduced to ${ }^{16}$

Condon and Slater showed that the $n$-electron integral $\left\langle D^{\prime}\right| \hat{B}|D\rangle$ can be reduced to sums of certain one- and two-electron integrals. The derivation of these Condon-Slater formulas uses the determinant expression of Prob. 8.22 together with the orthonormality of the spin-orbitals. (See Parr, pp. 23-27 for the derivation.) Table 11.3 gives the Condon-Slater formulas.

In Table 11.3, each matrix element of $\hat{g}_{12}$ involves summation over the spin coordinates of electrons 1 and 2 and integration over the full range of the spatial coordinates of electrons 1 and 2 . Each matrix element of $\hat{f}_{1}$ involves summation over the spin coordinate of electron 1 and integration over its spatial coordinates. The variables in the sums and definite integrals are dummy variables.

TABLE 11.3 The Condon-Slater Rules

| $D$ and $D^{\prime}$ differ by | $\left\langle D^{\prime}\right\| \sum_{i=1}^{n} \hat{f}_{i}\|D\rangle$ | $\left\langle D^{\prime}\right\| \sum_{i=1}^{n-1} \sum_{j>1} \hat{g}_{i j}\|D\rangle$ |
| :---: | :---: | :---: |
| no spin-orbitals | $\sum_{i=1}^{n}\left\langle u_{i}(1)\right\| \hat{f}_{1}\left\|u_{i}(1)\right\rangle$ | $\begin{aligned} \sum_{i=1}^{n-1} \sum_{j>1}[ & \left\langle u_{i}(1) u_{j}(2)\right\| \hat{g}_{12}\left\|u_{i}(1) u_{j}(2)\right\rangle \\ & \left.-\left\langle u_{i}(1) u_{j}(2)\right\| \hat{g}_{12}\left\|u_{j}(1) u_{i}(2)\right\rangle\right] \end{aligned}$ |
| one spin-orbital $u_{n}^{\prime} \neq u_{n}$ | $\left\langle u_{n}^{\prime}(1)\right\| \hat{f}_{1}\left\|u_{n}(1)\right\rangle$ | $\begin{aligned} & \sum_{j=1}^{n-1}\left[\left\langle u_{n}^{\prime}(1) u_{j}(2)\right\| \hat{g}_{12}\left\|u_{n}(1) u_{j}(2)\right\rangle\right. \\ & \left.-\left\langle u_{n}^{\prime}(1) u_{j}(2)\right\| \hat{g}_{12}\left\|u_{j}(1) u_{n}(2)\right\rangle\right] \end{aligned}$ |
| two spin-orbitals | 0 | $\left\langle u_{n}^{\prime}(1) u_{n-1}^{\prime}(2)\right\| \hat{g}_{12}\left\|u_{n}(1) u_{n-1}(2)\right\rangle$ |
| $u_{n}^{\prime} \neq u_{n}, u_{n-1}^{\prime} \neq u_{n-1}$ |  | $-\left\langle u_{n}^{\prime}(1) u_{n-1}^{\prime}(2)\right\| \hat{g}_{12}\left\|u_{n-1}(1) u_{n}(2)\right\rangle$ |
| three or more spin-orbitals | 0 | 0 |

## TABLE 11.3 The Condon-Slater Rules

| $D$ and $D^{\prime}$ differ by | $\left\langle D^{\prime}\right\| \sum_{i=1}^{n} \hat{f_{i}}\|D\rangle$ | $\left\langle D^{\prime}\right\| \sum_{i=1}^{n-1} \sum_{j>1} \hat{g}_{i j}\|D\rangle$ |
| :---: | :---: | :---: |
| no spin-orbitals | $\sum_{i=1}^{n}\left\langle u_{i}(1)\right\| \hat{f}_{1}\left\|u_{i}(1)\right\rangle$ | $\sum_{i=1}^{n-1} \sum_{j>1}\left[\left\langle u_{i}(1) u_{j}(2)\right\| \hat{g}_{12}\left\|u_{i}(1) u_{j}(2)\right\rangle\right.$ |
|  |  | $\left.-\left\langle u_{i}(1) u_{j}(2)\right\| \hat{g}_{12}\left\|u_{j}(1) u_{i}(2)\right\rangle\right]$ |
| one spin-orbital | $\left\langle u_{n}^{\prime}(1)\right\| \hat{f}_{1}\left\|u_{n}(1)\right\rangle$ | $\sum_{j=1}^{n-1}\left[\left\langle u_{n}^{\prime}(1) u_{j}(2)\right\| \hat{g}_{12}\left\|u_{n}(1) u_{j}(2)\right\rangle\right.$ |
| $u_{n}^{\prime} \neq u_{n}$ |  | $\left.-\left\langle u_{n}^{\prime}(1) u_{j}(2)\right\| \hat{g}_{12}\left\|u_{j}(1) u_{n}(2)\right\rangle\right]$ |
| two spin-orbitals | 0 | $\left\langle u_{n}^{\prime}(1) u_{n-1}^{\prime}(2)\right\| \hat{g}_{12}\left\|u_{n}(1) u_{n-1}(2)\right\rangle$ |
| $u_{n}^{\prime} \neq u_{n}, u_{n-1}^{\prime} \neq u_{n-1}$ |  | $-\left\langle u_{n}^{\prime}(1) u_{n-1}^{\prime}(2)\right\| \hat{g}_{12}\left\|u_{n-1}(1) u_{n}(2)\right\rangle$ |
| three or more spin-orbitals | 0 | 0 |

no mismatches = leading term of identical determinants
$D$ and $D^{\prime}$ differ by

$$
\langle p| \sum_{i} \hat{\|}|\rho\rangle
$$

no spin-orbitals

$$
\sum_{i=1}^{n}\left\langle u_{i}(1)\right| \hat{f}_{1}\left|u_{i}(1)\right\rangle
$$

leading terms of identical determinants
one spin-orbital
$\left\langle u_{n}^{\prime}(1)\right| \hat{f}_{1}\left|u_{n}(1)\right\rangle$
$u_{n}^{\prime} \neq u_{n}$
two spin-orbitals
$u_{n}^{\prime} \neq u_{n}, u_{n-1}^{\prime} \neq u_{n-1}$
three or more
spin-orbitals

1 - electron operators give $2 h_{A A}+2 h_{B B}+2 h_{C C}$
where $h_{A A}$ is:

$$
\left.\mathrm{h}_{\mathrm{AA}}=<\mathrm{A}(1)\left|-\frac{1}{2} \nabla_{1}^{2}-\sum_{\mu} \frac{Z_{1 \mu}}{r_{1 \mu}}\right| \mathrm{A}(1)\right\rangle
$$


= the kinetic energy and potential energy of attraction to all nuclei for an electron in orbital $A$



## $D$ and $D^{\prime}$ differ by

no spin-orbitals
leading terms of identical determinants
one spin-orbital
$u_{n}^{\prime} \neq u_{n}$
two spin-orbitals
$u_{n}^{\prime} \neq u_{n}, u_{n-1}^{\prime} \neq u_{n-1}$

$$
\begin{aligned}
& \sum_{j=1}^{n-1}\left[\left\langle u_{n}^{\prime}(1) u_{j}(2)\right| \hat{g}_{12}\left|u_{n}(1) u_{j}(2)\right\rangle\right. \\
& \left.-\left\langle u_{n}^{\prime}(1) u_{j}(2)\right| \hat{g}_{12}\left|u_{j}(1) u_{n}(2)\right\rangle\right] \\
& \left\langle u_{n}^{\prime}(1) u_{n-1}^{\prime}(2)\right| \hat{g}_{12}\left|u_{n}(1) u_{n-1}(2)\right\rangle \\
& \quad-\left\langle u_{n}^{\prime}(1) u_{n-1}^{\prime}(2)\right| \hat{g}_{12}\left|u_{n-1}(1) u_{n}(2)\right\rangle
\end{aligned}
$$

three or more
spin-orbitals


2 - electron operators give
$\mathrm{J}_{\mathrm{AA}}+\mathrm{J}_{\mathrm{BB}}+\mathrm{J}_{\mathrm{CC}}+4 \mathrm{~J}_{\mathrm{BA}}+4 \mathrm{~J}_{\mathrm{AC}}+4 \mathrm{~J}_{\mathrm{BC}}-2 \mathrm{~K}_{\mathrm{BA}}-2 \mathrm{~K}_{\mathrm{CA}}-2 \mathrm{~K}_{\mathrm{BC}}$
Electron repulsion is reduced between electron with same $m_{s}$ as seen for $\mathrm{He}(1 \mathrm{~s}, 2 \mathrm{~s})$

